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## Some Studies On Exploring Potential Of Zeolites As A Drug Delivery Carrier.

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### ABSTRACT

In this work we used a combination of different techniques to investigate the adsorption properties of various prototype drugs (Atorvastatin Calcium, Metoprolol Succinate & Nabumetone) by zeolite type 4A for potential use as a drug carrier. Zeolites are inorganic crystalline aluminosilicates that have a large internal surface area and a porous structure at the nano and microscales. They can be used as pharmaceutical carrier systems to enhance the dissolution of medications that have a low water solubility by incorporating them. Here, we used SEM, PXRD, thermo gravimetric analysis (TGA), and UV-vis spectroscopy to investigate the surface properties and morphology of zeolite type 4A before and after loading of the antihyperlipidemic, antihypertensive, and NSAID (Non-Steroidal Anti-Inflammatory Drug) agent. The findings are utilized to evaluate the zeolite type 4A's loading efficiency and its post-loading structural stability. CCD (Central Composite Design) was utilised for optimization which concludes that drug to carrier ratio of 1:2 while poloxamer concentration of 0.10%w/v found to be optimum. Test formulations showed sustained drug release profile of upto 10 hours when compared with marketed ones.

**Keywords:** Carrier, Zeolite 4A, Central Composite Design, Drug release

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## INTRODUCTION

When compared to conventional pharmacological treatments, drug delivery systems have demonstrated increased efficiency; yet, certain obstacles and restrictions still exist. Developing innovative drug delivery methods, like those based on inorganic molecular carriers, might be very helpful in resolving these issues [1,2]. Zeolite structures are among the inorganic drug carriers that have been the subject of many investigations because of their special qualities, which include high adsorption capacity and specific surface area, water solubility, strong biocompatibility, and stability in biological contexts [12,13].

Zeolites, which have a high internal surface area and a porous structure on the micro-and nanoscale, are inorganic crystalline aluminosilicates that are widely used as ion exchange materials, molecular sieves, and adsorbents in various applications. Zeolites have lately shown great promise in the medical and pharmaceutical domains because of their bioactive qualities and strong durability in biological sites [2,7,11].

In nature, aluminosilicate minerals are the form of many zeolites, and over 200 synthetic zeolites can be created in a lab. The structure of each zeolite framework is distinct and is often composed of silicon, aluminium, and oxygen atoms. Connectivity between  $[\text{SiO}_4]^{4-}$  and  $[\text{AlO}_4]^{5-}$  units arranges the building blocks in a periodic pattern to create cages and channels that give the material its distinctive microporous structure. Zeolites' distinct set of characteristics has led to a variety of medical uses for them, including as the transport and encapsulation of medications with antibacterial, anticancer, and antibiotic qualities [16,17]. With regard to drug delivery, a patient receiving cancer therapy, for example, could receive zeolitic structures loaded with drugs without losing the unique pharmacological effects of the medications [18].

This study examines the capacity of synthetic zeolite 4A, a  $\text{Na}^+$  exchanged variant of zeolite LTA (Figures 1 & 2), to bind the prototype medications, which are widely recognized for their medicinal properties. With an effective pore opening of approximately 4 Å and  $\text{Na}^+$  ion acting as a coordinating cation in its structure, zeolite 4A provides a more consistent and reliable drug delivery system. [19,20].

## MATERIALS AND METHODS

Zeolite 4A and Pluronic F-127 (Poloxamer) were obtained from Sigma Aldrich. Atorvastatin Calcium, Metoprolol Succinate & Nabumetone were obtained as free gift sample from Accent Pharma, Cipla & Micro Labs Ltd India respectively. Every excipient and solvent used was of analytical grade.

### Method of Formulating Drug Loaded Zeolite

Prior to formulation, carrier zeolite 4A is activated by placing it in muffle furnace with the help of guch crucible. Generally solvent evaporation method is employed. This activated zeolite along with Atorvastatin Calcium and poloxamer were mixed in dichloromethane which then evaporated to obtain crude powder. Finally, capsule was filled with other ingredients like magnesium stearate & microcrystalline cellulose. Same process is used for remaining two drugs i.e. Metoprolol Succinate & Nabumetone [17,18].

### Preformulation Studies

The physicochemical characteristics of the drug and excipients are described by preformulation studies in order to create a stable, dependable, and safe dosage form. Preformulation studies include organoleptic properties, solubility, UV spectroscopy, melting point determination & compatibility studies (FTIR) [3-6,8,9].

### Organoleptic properties

Drugs namely Atorvastatin Calcium (AC), Metoprolol Succinate (MS) & Nabumetone (NU) were observed for its colour, odour and appearance by visual observation.

## UV-Visible spectroscopy

Calibration curves for all 3 drugs were performed in PBS pH 6.8 to obtain the equation and R<sup>2</sup> value.

## Preparation of stock solution

Stock solution was prepared for concentration of 100 and 1000 µg/ml to find out fixed wavelength where the maximum peak is obtained [14].

## Procedure for calibration curve of AC, MS & NU

After getting fixed wavelength different dilutions (i.e. 5, 10, 15, 20, 25 µg/ml) were prepared from the stock solution of 100 µg/ml in to a 10ml volumetric flasks. The final volume was made by the PBS pH 6.8 solvent used and absorbance was measured at 245nm, 255nm & 340nm for AC, MS & NU respectively by using Shimadzu UV-1800 UV-visible spectrophotometer with quartz cells of 10mm path length [13-15].

## Experimental Work

### Preliminary Study

This screening is carried out to see how different ratios of drug to carrier affected the formulation of drug loaded zeolite as shown in Table 3 & 7. These trial batches are evaluated for different parameters like particle size, zeta potential, in-vitro dissolution study and loading efficacy [17-19]. On the basis of this parameters the suitable ratio of drug to carrier is selected for further formulation. From this it was concluded that 1:2 or 1:5 could be better for optimization so taken forward [38-40].

### Formulation of Drug Loaded Zeolite by Solvent Evaporation Method [10,18]

- API i.e. Atorvastatin Calcium (40mg), Activated Zeolite 4A (80mg) and Poloxamer F-127 (30mg) were dissolved in dichloromethane.
- The above mixture was stirred for few minutes & kept aside for evaporation by covering with fine sieve on beaker.
- The crude product was obtained by scrapping and allowed to dry in oven to remove volatile content (if any).

Same process has been applied for another two API namely Metoprolol Succinate & Nabumetone

### Formulation and optimization of Solid dispersion

The optimization of solid dispersion was performed by central composite design [25]. A 150 mg solid dispersion was prepared by solvent evaporation method utilizing dichloromethane as solvent [10]. Ingredients used in formulation were Zeolite-4A as a carrier and Pluronic F-127 as stabiliser. Microcrystalline cellulose & magnesium stearate were other excipients for capsule filling (Shown in Table 8 & 9).

### Name of design

Central Composite Design (CCD)

### Independent variables

- A: API to Carrier Ratio
- B: Concentration of Poloxamer F-127 (%)

### Dependent variables

- Y1: Particle Size,

- Y2: Zeta Potential &
- Y3: % CDR

### Evaluation of batches

By applying Central Composite design about 13 experimental runs were obtained and particle size, zeta potential & % CDR were taken as response variable [20-23].

### Particle Size

- 3 formulations from each run were evaluated for particle size and the result was noted down.
- 10mg powder of product was utilized for this test.

### Zeta Potential

- About 3 formulations of each drug containing zeolite of each run were evaluated for this test
- Zeta Potential was noted for each run

### % CDR

- Dissolution Medium: 900 ml of Phosphate Buffer Solution (pH 6.8)
- Temperature:  $37 \pm 1^\circ\text{C}$
- Speed: 50 rpm
- Dissolution apparatus: USP Type-I (Basket Apparatus)

### Selection and detail evaluation of optimized batch physical mixture

From the 13 runs the optimized batch was selected and the composite mixture of the optimized batch was further evaluated [25].

### Particle Size

This was determined by using Horiba Analyser Instrument. 10mg of product was dissolved in 10 ml of solvent. From this 1 ml of formulation was diluted with 10 ml distilled water and was poured into the polystyrene cuvette which was placed in thermostatic chamber maintained at  $25^\circ\text{C}$ . Detection was carried out at scattering angle of  $90^\circ$  [20]

### Zeta Potential

Zeta potential of the formulation was determined by dynamic light scattering using particle size analyzer (Horiba). The samples were diluted with a ratio of 1:10 (v/v) with distilled water [21].

### In-Vitro Dissolution Study

The dissolution study of the prepared drug loaded zeolite was performed by using USP type-I dissolution apparatus i.e. Basket. Capsule (filled with 100 mg of product) was placed in dissolution vessel and the dissolution study was carried out at  $37 \pm 1^\circ\text{C}$  using 900 ml of phosphate buffer solution (pH 6.8) as the dissolution medium. 5 ml of sample was withdrawn periodically at 15 min, 30 min, 60 min, like that up to 10 hours and each sample is replaced with equal volume of fresh dissolution medium in order to maintain sink condition [22-23]. Samples were analyzed by UV spectrophotometer at 245 nm, 255 nm and 340 nm for AC, MS & NU release respectively (Table 11, 12 & 13)

### PXRD (Powder X-ray Diffraction)

X-ray diffraction patterns for the empty and drug loaded zeolite were obtained with a Rikagu (Ultima IV X-ray diffractometer, SPPU, Pune, India). All samples were analyzed in powder form [29]. With a counting time of two seconds, diffraction data were gathered in the  $2\theta$  range, which spans  $5^\circ$  to  $80^\circ$  and corresponds to a scan speed of 10 degrees per minute.

### **TGA (Thermo Gravimetric Analysis)**

Drug content in the Zeolite 4A formulation was additionally quantified with thermogravimetric analysis using a TA Instruments Trios V4.4.0.41128. Samples (Drug with Zeolite 4A) were analyzed in N<sub>2</sub> atmosphere at a heating rate of 10°C/min & in the temperature range from 0°C to 800°C [30].

### **SEM (Scanning Electron Microscopy)**

The shape and surface morphological properties of empty carrier i.e. zeolite 4A and drug (Atorvastatin Calcium, Metoprolol Succinate & Nabumetone) loaded zeolite were examined by Scanning Electron Microscopy (SPPU JEOL JSM 6360A, India). Dry zeolite 4A & product powder is sprinkled lightly on double adhesive tape which is stuck to an aluminium stub which was then placed in the SEM chamber and scanned. After that the photomicrographs taken at magnification 300x, 1000x & 10000x [26,27].

### **FTIR (Fourier Transform Infrared Spectroscopy)**

The IR study of drug loaded zeolite was performed using potassium bromide (KBr) as blank. The samples to be analyzed and KBr were previously dried in the oven for 30 min and mixed thoroughly in 1:100 (sample : KBr) ratio in a glass mortar. These samples were then placed in a sample holder and scans were obtained at a resolution of 2cm<sup>-1</sup> from 400 to 4000 cm<sup>-1</sup> [35-37]

### **Loading Efficiency**

Drug content in the Zeolite formulations was quantified using the extraction method described below. 10mg of sample was dispersed in 10 mL of PBS pH 6.8 and was magnetically stirred for 24h at 37°C. Dispersions centrifuged for 15 min, at 4500 rpm, supernatants filtered through a whatman filter of 0.45µm pore size and drug content was quantified with UV spectroscopy (UV Jasco V-630) at 245 nm, 255 nm and 340 nm respectively [29,30] (Table 10)

Drug loading efficiency was calculated according to following equation:

$$\text{Drug Loading Efficiency (\%)} = \frac{\text{Weight of drug in the particles} \times 100}{\text{Weight of particles}}$$

### **Drug content**

The powder of drug loaded zeolite equivalent to 10 mg drug was dissolved in 100ml of water. One ml of above solution was withdrawn and volume was made up to 10 ml [28]. The absorbance was measured at 245, 255 & 340 nm against the PBS pH 6.8 as blank solution by using UV Spectrophotometer (Table 10).

### **Docking study**

For this Atorvastatin calcium complexed with zeolite was taken as ligand while SARS CoV-2 was taken as receptor. The structures were generated initially in the 2D form utilizing Chemdraw® Ultra software. LigPrep program of the Maestro 9.1 software was employed in preparing the ligands for molecular docking study. The receptor co-crystal structures of COVID-19 main protease in complex with an inhibitor N3 (PDB ID: 6LU7) was obtained from the RCSB Protein Data Bank. The protein structures were prepared using Protein Preparation Wizard of Maestro 9.1 [31].

### **Stability study**

On the basis of ICH-Q1A guidelines, the stability testing of optimized drug loaded zeolite batch formulation was performed [32-34]. The optimized samples were kept in air tight amber coloured glass container for 1 month to perform accelerated stability study. Factors like appearance, average weight & moisture content were evaluated after 1 month of time period (Table 15).

## RESULTS AND DISCUSSION

### Preformulation studies

#### Identification of drug

#### Organoleptic properties, solubility and melting point of AC, MS & NU

The organoleptic properties, solubility & melting point are summarized in (Table 1).

#### UV-Visible Spectrophotometry

Calibration curve for AC, MS & NU were performed in PBS pH 6.8. According to Beer-Lambert's law, the absorption maximum was observed at 245nm for AC, 255nm for MS & 340nm for NU. Absorbance data and calibration curves for AC, MS & NU in PBS pH 6.8 are given in following (Table 2 and Figure 3,4,5).

### Result & Discussion For Formulation And Evaluation of Drug Loaded Zeolite

#### Evaluation of batches

13 runs obtained by applying central composite design were further evaluated for their particle size, zeta potential & % CDR. Run 1 (Batch Code F1) results were found to be relatively good as compare to other runs as shown in (Table 9). From this run 1 was found to be the optimized batch.

#### Particle size distribution

Particle size of optimized drug loaded zeolite formulations was found to be 90.9 nm (Atorvastatin Calcium), 70.8 nm (Metoprolol Succinate) & 122.8 nm (Nabumetone) respectively i.e. size is in the range specified for nanocarriers (Shown in Table 10 & Figure 7,8,9).

#### Zeta Potential

Zeta potential of the optimized drug loaded zeolite batch was found to be -23.7 mV (AC), - 16.1 mV (MS) & -11.3 mV (NU) respectively. The values were found in the standard range hence it can be concluded that the optimized batch is stable (Shown in Table 10 & Figure 10,11,12).

#### In vitro drug release kinetics for optimized batch of solid dispersion (Run 1)

The potential of solid dispersion to modulate drug release is shown by their in vitro release kinetics, which is an important metric to take into account when evaluating the products' safety, effectiveness, and quality. (Table 14) shows the drug release kinetics [41-49].

#### Comparative study of formulated capsule with the marketed formulation

For marketed comparison, Ecosprin 40/75 AV (USV Pvt Ltd) contains 40mg Atorvastatin Calcium & Metolar XR 50 (Cipla) contains 50mg Metoprolol Succinate were taken as reference brands for drug release comparison. It has shown drug release occurred upto 8 hours in marketed ones while test formulations showed upto 9-10 hours (Shown in Table-11,12 & Figure 16 & 17).

#### FTIR (Fourier Transfer Infrared Spectrophotometer)

FTIR spectra of pure drug and drug loaded zeolite were evaluated (Shown in Table 4,5,6 & Figure 6). The FTIR study for both pure drug & solid dispersion was carried out and the observed peaks were noted. The peaks of the drug were retained in the solid dispersion formulation as in pure drug, which indicates drug and excipients were found to be compatible with each other [35- 37].

### **PXRD Study**

Figure 18 displays all of the distinctive peaks for both the drug and zeolite 4A. It is evident from XRD that the framework has not altered. Peak intensity decreases after drug adsorption indicated a slight drop in zeolite crystallinity.

### **TGA Study**

Thermogravimetric analysis (TGA) profile of product is shown in figure 20. It can be observed a first weight loss in the temperature range 50-100°C corresponding mainly to volatile content & water desorption and a second step between 100 and 400°C attributed to the decomposition and combustion of the organic fraction (drug). From the weight loss calculated for both steps, about 42-77% w/w drug presence on the surface of zeolite. This may be due to the interaction between silanol groups and carbonyl/hydroxyl groups of AC, MS & NU.

### **Scanning Electron Microscopy (SEM)**

Morphological characteristics of empty zeolite & drug loaded zeolite were observed using SEM analysis. SEM photomicrograph of optimized formulation at magnification 300x & 1000x as shown in figure 19. On the basis of SEM results it can be evident that drug has been loaded on zeolitic surface and average size was found between 10-50 microns.

### **Docking Study**

As per the docking study & binding energy (Shown Figure 21 & Table 16), it can be evident that the ligand Atorvastatin has higher affinity towards SARS-CoV 2 receptor than complex. So from this it is clear that atorvastatin has more than 1 target i.e. HMG-CoA reductase enzyme & SARS-CoV 2 receptor.

### **Stability studies**

Stability testing under expedited circumstances the optimized solid dispersions were demonstrated to be stable over a one-month period at 40°C±2°C/75%RH±5%RH relative humidity by the lack of substantial alterations in their physical characteristics and in the following parameters (Figure 22)

## **CONCLUSION**

Solid dispersions of drug loaded zeolite proved to be one of the successful approaches to overcome the problem associated with conventional drug delivery (like low bioavailability & frequent dosing); also which enhances patient compliance and sustained drug release profile. Atorvastatin Calcium, Metoprolol Succinate & Nabumetone were loaded onto the zeolitic surface for enhanced drug release. Central composite design was employed for optimization study. Particle size, zeta potential & % drug release were crucial response parameters. Results are encouraging with respect to zeolite solid dispersion. Optimized batch showed better & sustained drug release as compared to marketed formulations. The optimized batch's stability study yielded satisfactory results as well. Main building blocks adopted in this work are molecular weight, hydrophobicity, functional groups & drug:carrier ratio; these could be explored in depth in addition with others to get more insights in zeolites. Study has wide future scope with respect to methods of preparation, incorporation & physicochemical properties of drug such as solubility, polymorphism, etc. Potent drugs with complex metabolism can be tried to deliver using this carrier.

## **ACKNOWLEDGEMENTS**

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**Table 1: Preformulation Study of AC, MS & NU**

Parameter	Observation		
	AC	MS	NU
Physical description	Solid	Solid	Solid
Structure	Crystalline	Crystalline	Crystalline
Colour	White	White	White
Odour	Odourless	Odourless	Odourless
Taste	Unpleasant taste	Unpleasant taste	Unpleasant taste
Solubility	Soluble in organic solvent (Dichloromethane)	Soluble in organic solvent (Dichloromethane)	Soluble in organic solvent (Dichloromethane)
Melting point	164±2°C	138±2°C	80±2°C

**Table 2: Concentration vs Absorbance values for development of standard curve in PBS pH 6.8**

Conc. (µg/ml)	AC i) Abs (245 nm)	MS ii) Abs (255 nm)	NU iii) Abs (340 nm)
5	0.38	0.14	0.09
10	0.65	0.28	0.18
15	0.98	0.41	0.26
20	1.22	0.51	0.35
25	1.52	0.68	0.43

**Table 3: Preliminary trial batches for optimum drug to carrier ratio**

Batch Code	Drug to Carrier Ratio
B1	1:1
B2	1:2
B3	1:5
B4	1:10

**Table 4: FTIR Spectra Elucidation of AC loaded zeolite**

Functional Group	Observed Value (cm-1)	Reference Value (cm-1)
Al-O	716	700-800
Si-O	946	900-1000
C=O	1665	1650-1800
C-H (Aliphatic)	2927	2850-2980
C-H (Aromatic)	3067	3000-3100
O-H	3630	3500-3700

**Table 5: FTIR Spectra Elucidation of MS loaded zeolite**

Functional Group	Observed Value (cm-1)	Reference Value (cm-1)
Al-O	808	700-800
Si-O	966	900-1000
C-H (Aliphatic)	2973	2850-2980
N-H	3273	3100-3300
O-H	3584	3500-3700



**Table 6: FTIR Spectra Elucidation of NU loaded zeolite**

Functional Group	Observed Value (cm-1)	Reference Value (cm-1)
Al-O	716	700-800
Si-O	946	900-1000
C=O	1767	1650-1800
C-H (Aliphatic)	2896	2850-2960
C-H (Aromatic)	3053	3000-3100

**Table 7: Preliminary trial batches for selection of drug to carrier ratio**

Preliminary Study					
Batch Code	Drug to Carrier Ratio	Particle Size (nm)	Zeta Potential (mV)	% CDR	Loading Efficiency (%)
B1	01:01	350.5	-10.9	90	50
B2	01:02	90.9	-23.7	32	80
B3	01:05	283.1	-36	23	70
B4	01:10	400.8	-15.6	10	45

**Table 8: Capsule filling formula**

Sr No	Ingredient	Amount (mg)
1.	Crystallized Powder (Atorvastatin-Zeolite Dispersion)	80
2.	Magnesium Stearate	10
3.	Microcrystalline Cellulose	10

**Table 9: Composition of different batches of Atorvastatin Calcium loaded zeolite using CCD**

Sr. No.	Batch Code	A: API to Carrier Ratio	B: Conc. of Poloxamer F-127 (%)
1	F1	2	0.1
2	F2	5	0.1
3	F3	2	0.25
4	F4	5	0.25
5	F5	1.37868	0.175
6	F6	5.62132	0.175
7	F7	3.5	0.068934
8	F8	3.5	0.281066
9	F9	3.5	0.175
10	F10	3.5	0.175
11	F11	3.5	0.175
12	F12	3.5	0.175
13	F13	3.5	0.175

**Table 10: Summary of evaluated parameters of optimized batch F1 with SD (Standard Deviation)**

Sr. No.	Formulation	Particle Size (nm) Mean±SD	Zeta Potential (mV) Mean±SD	Drug Loading (%) Mean±SD	Drug Content (%) Mean±SD
1	AC+Zeolite 4A	92.63±2.12	-23.4±0.80	80.5±1.80	76.33 ±1.53
2	MS+Zeolite 4A	72.93±2.1	-15.56±0.61	72.5±0.5	73.9 ± 1.1
3	NU+Zeolite 4A	122.37±2.08	-12.16±0.70	71.3±1.13	72.13 ± 0.71

**Table 11: In-vitro drug release of test vs marketed formulation Ecosprin 75/40 AV**

Time (Hr)	% CDR (Zeolite+AC)	% CDR Marketed (Ecosprin 75/40 AV)
0.25	0.8	8.2
0.5	2.9	15.0
1	9.3	22.9
2	19.7	31.0
3	26.2	41.7
4	37.7	51.8
5	59.8	63.0
6	78.1	77.0
7	88.4	93.0
8	94.9	101.1
9	97.4	-
10	100.3	-

**Table 12: In-vitro drug release of test vs marketed formulation Metolar XR 50**

Time (Hr)	% CDR (Zeo+MS)	% CDR Marketed (Metolar XR 50)
0.25	3.4	13.3
0.5	10.9	26.5
1	18.5	27.6
2	21.9	29.8
3	29.5	33.3
4	41.3	45.2
5	53.8	57.2
6	86.4	71.1
7	98.2	87.0
8	101.6	97.9
9	-	101.3

**Table 13: In-vitro drug release of NU loaded zeolite in PBS pH 6.8**

Time (Hr)	% CDR (NU) loaded Zeolite
0.25	6.6
0.5	18.5
1	26.5
2	39.7
3	47.6
4	62.2
5	78.1
6	88.7
7	96.6
8	101.1

**Table 14: Drug Release Kinetics profile of drugs**

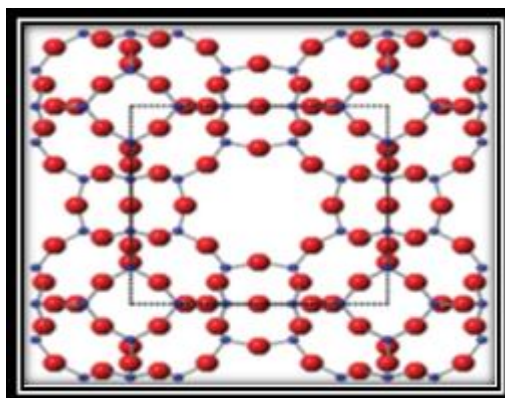
Batch Code	Model					Best Fit Model
	1 <sup>st</sup> Order	Zero Order	Higuchi Plot	Korsmeyer-Peppas Plot		
	R <sup>2</sup>	R <sup>2</sup>	R <sup>2</sup>	R <sup>2</sup>	n	
AC (F4)	0.8949	0.9693	0.9496	0.9846	1.29	KP Plot
MS (F4)	0.9999	0.9581	0.8879	0.9571	1.16	1 <sup>st</sup> Order
NU (F4)	0.8883	0.9333	0.9857	0.9757	0.727	Higuchi Plot

**Table 15: Storage conditions for stability study of drug loaded zeolite**

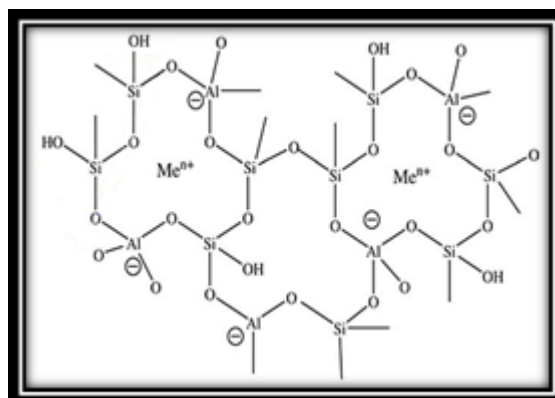
Study	Storage conditions	Time period
Accelerated	40°C ± 2°C & 75% RH ± 5% RH	1 month

**Table 16: Binding energies of various ligands on SARS CoV-2 receptor**

Sr. No.	Molecule	Binding Energy (Gscore)
1	Atorvastatin Calcium	-6.8482
2	Zeolite	-3.9174
3	Complex	-2.4633



**Figure 1: Zeolite Linde type A (LTA) framework Red: Oxygen atoms; Blue: Si/Al atoms**



**Figure 2: Chemical Structure of Zeolite**

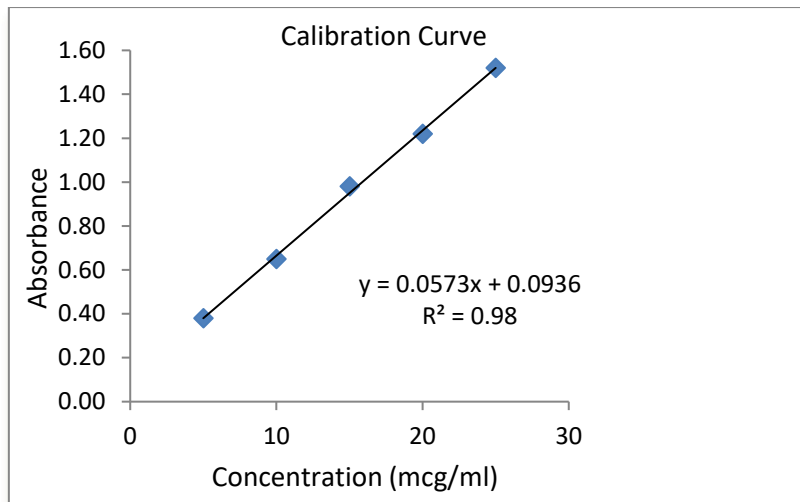


Figure 3: Calibration curve of AC in PBS pH 6.8

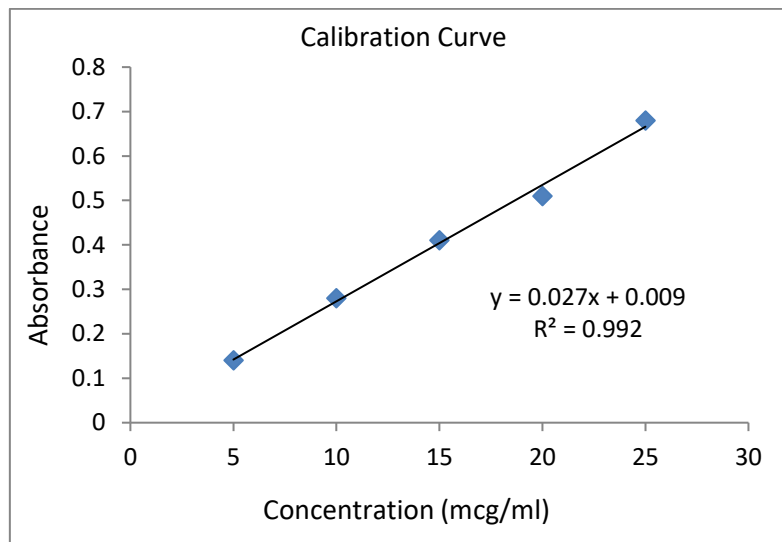


Figure 4: Calibration curve of MS in PBS pH 6.8

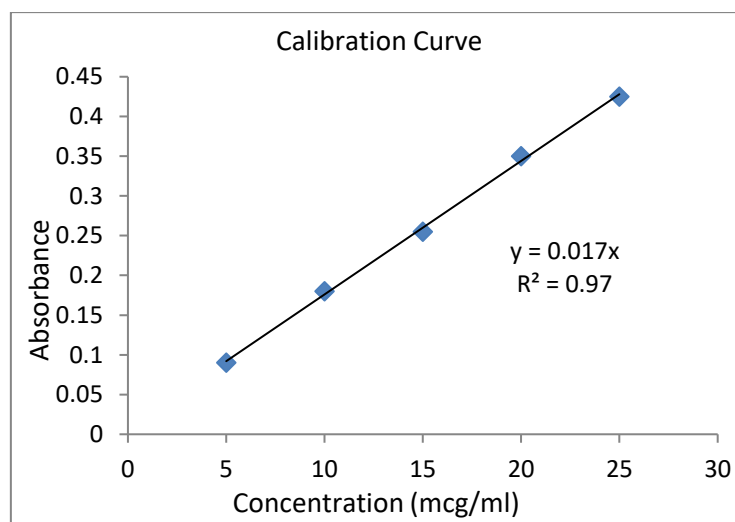


Figure 5: Calibration curve of NU in PBS pH 6.8

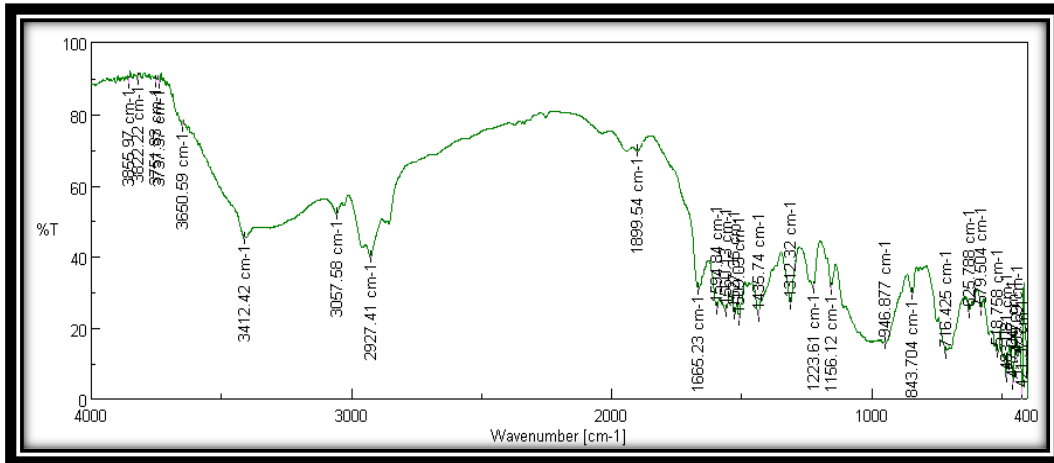


Figure 6: FTIR spectra of Atorvastatin Ca loaded Zeolite

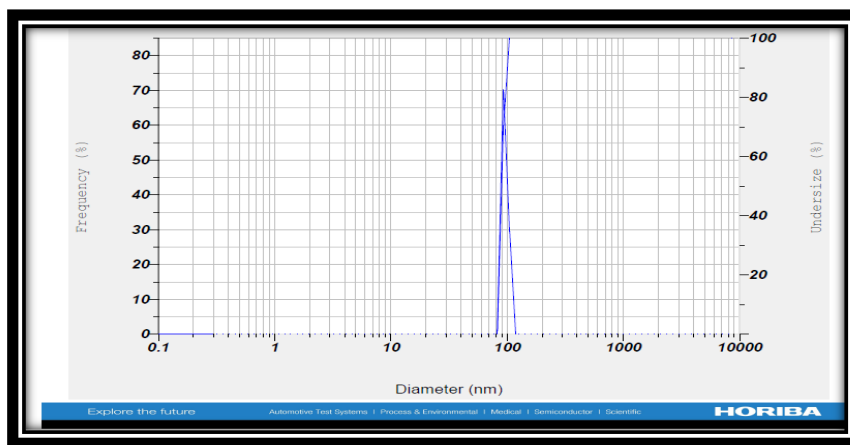


Figure 7: Particle size of optimized F1 AC loaded zeolite

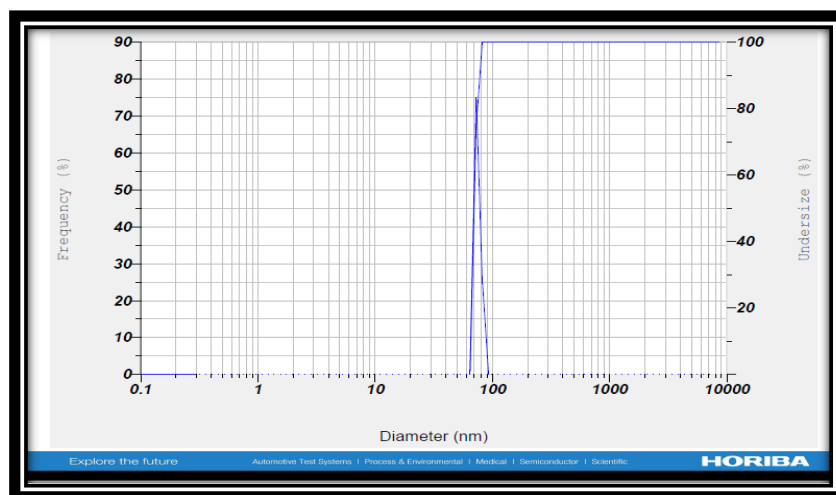


Figure 8: Particle size of optimized F1 MS loaded zeolite

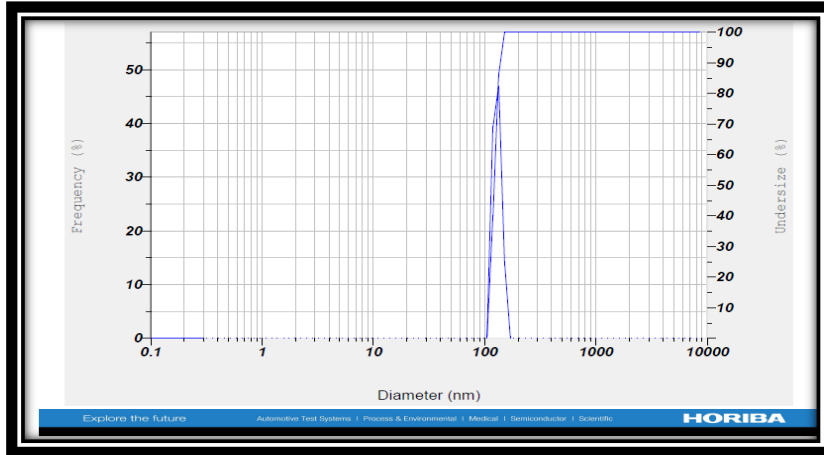


Figure 9: Particle size of optimized F1 NU loaded zeolite

**Calculation Results**

Peak No.	Zeta Potential	Electrophoretic Mobility
1	-23.7 mV	-0.000184 cm <sup>2</sup> /Vs
2	-- mV	-- cm <sup>2</sup> /Vs
3	-- mV	-- cm <sup>2</sup> /Vs

Zeta Potential (Mean) : -23.7 mV  
 Electrophoretic Mobility Mean : -0.000184 cm<sup>2</sup>/Vs

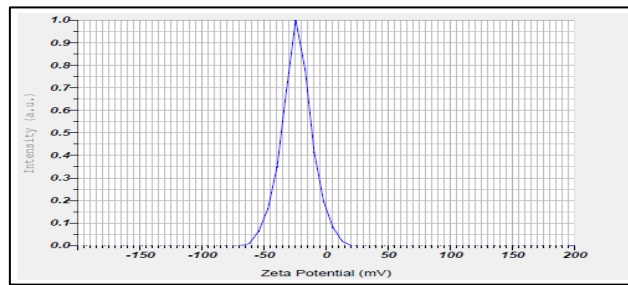


Figure 10: Zeta Potential of optimized F1 AC loaded zeolite

**Calculation Results**

Peak No.	Zeta Potential	Electrophoretic Mobility
1	-16.1 mV	-0.000125 cm <sup>2</sup> /Vs
2	-- mV	-- cm <sup>2</sup> /Vs
3	-- mV	-- cm <sup>2</sup> /Vs

Zeta Potential (Mean) : -16.1 mV  
 Electrophoretic Mobility Mean : -0.000125 cm<sup>2</sup>/Vs

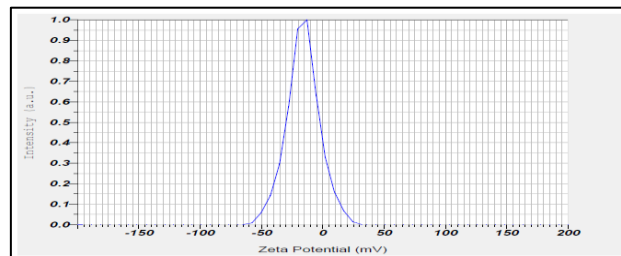


Figure 11: Zeta Potential of optimized F1 MS loaded zeolite

Calculation Results		
Peak No.	Zeta Potential	Electrophoretic Mobility
1	-11.3 mV	-0.000088 cm <sup>2</sup> /Vs
2	--- mV	--- cm <sup>2</sup> /Vs
3	--- mV	--- cm <sup>2</sup> /Vs
Zeta Potential (Mean)		: -11.3 mV
Electrophoretic Mobility Mean		: -0.000088 cm <sup>2</sup> /Vs

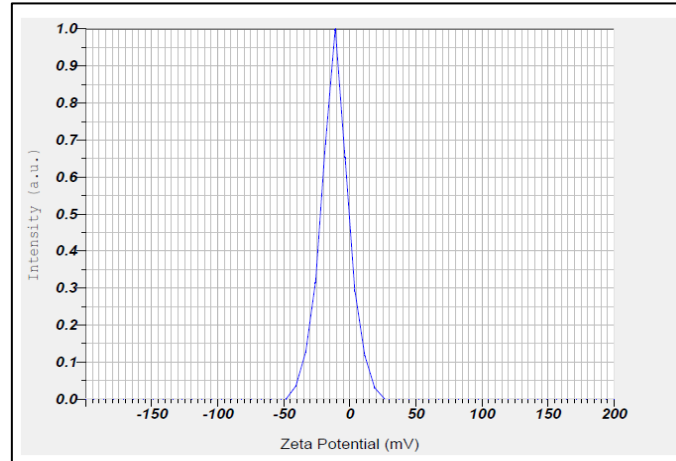


Figure 12: Zeta Potential of optimized F1 NU loaded zeolite

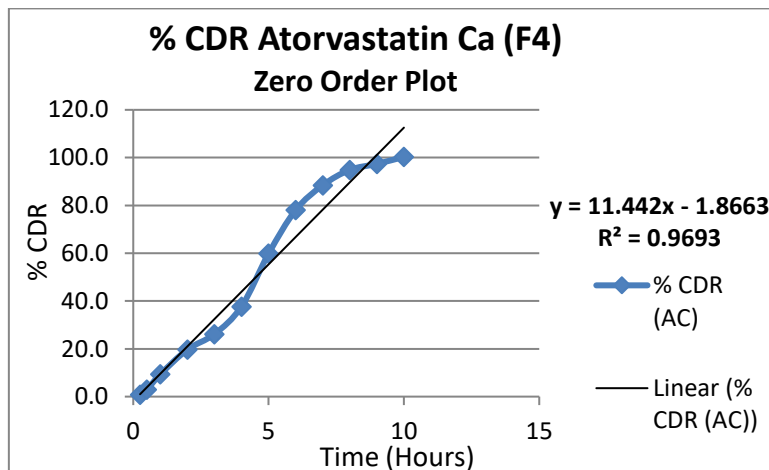


Figure 13: In-vitro drug release study of AC loaded zeolite in PBS pH 6.8

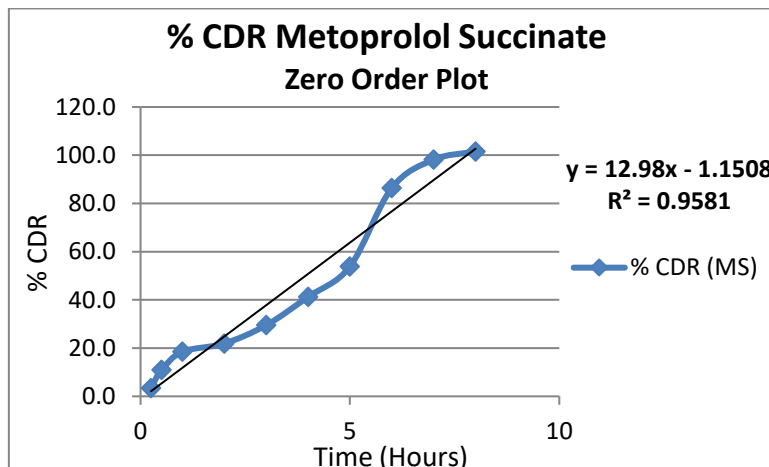


Figure 14: In-vitro drug release study of MS loaded zeolite in PBS pH 6.8



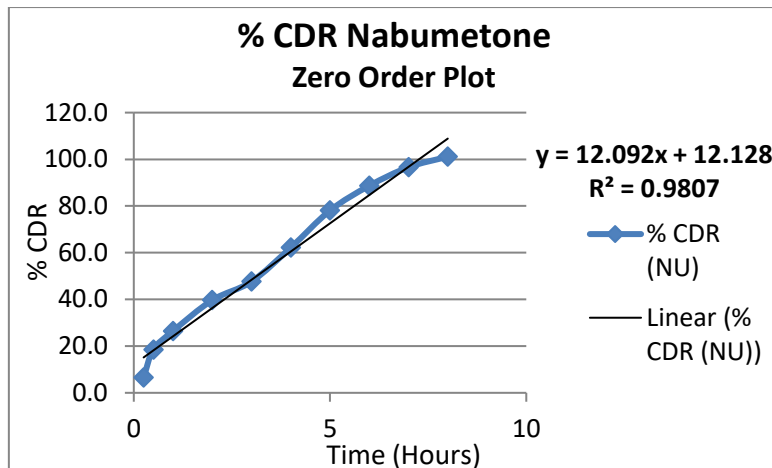


Figure 15: In-vitro drug release study of NU loaded zeolite in PBS pH 6.8

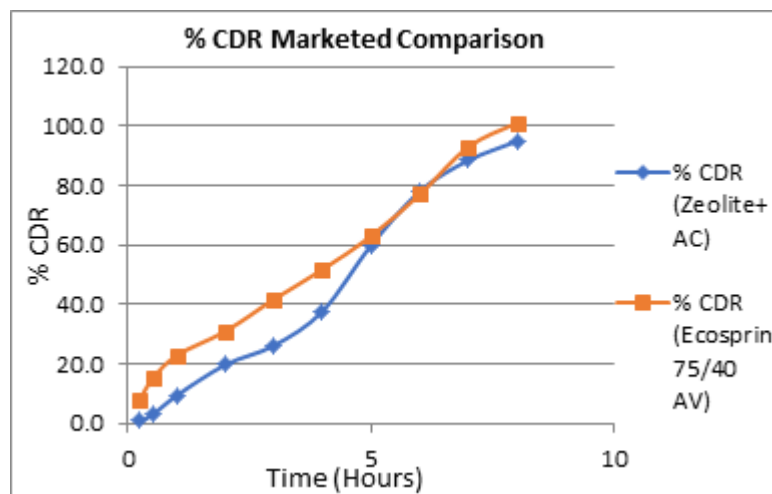


Figure 16: %CDR of Ecosprin 75/40 AV (Marketed) vs Test Formulation

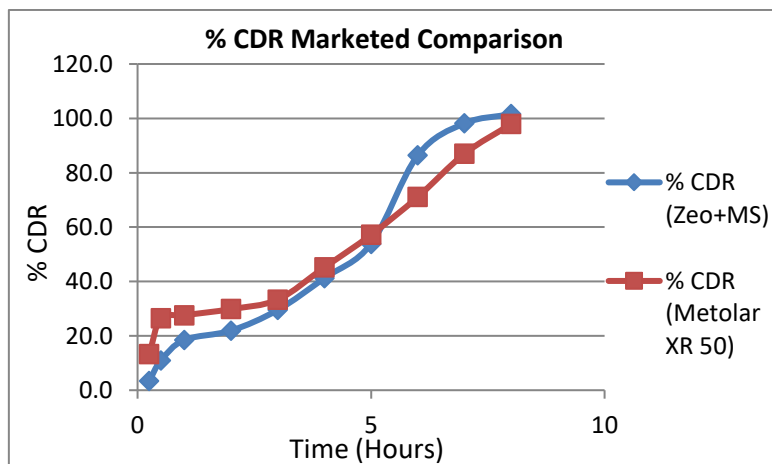


Figure 17: %CDR of Metolar XR 50 (Marketed) vs Test Formulation

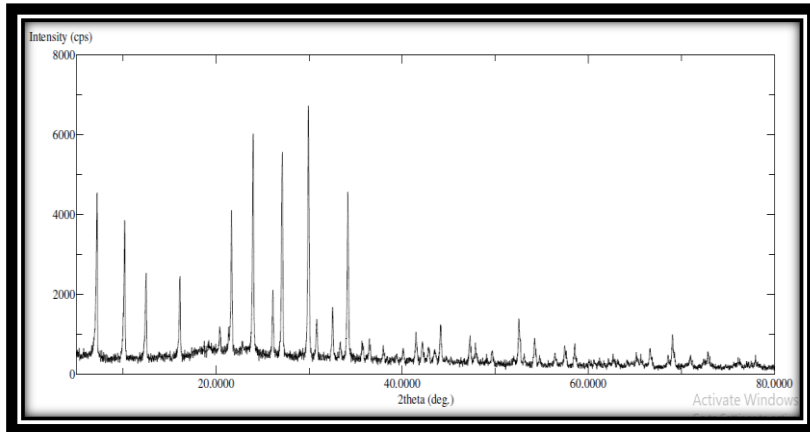


Figure 18: PXRD graph of AC loaded zeolite

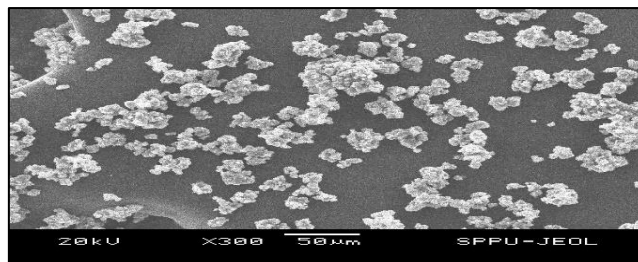


Figure 19: SEM image of AC loaded zeolite

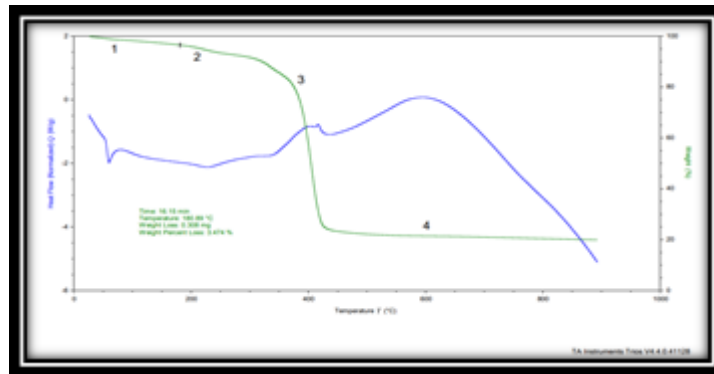


Figure 20: TGA of AC loaded zeolite

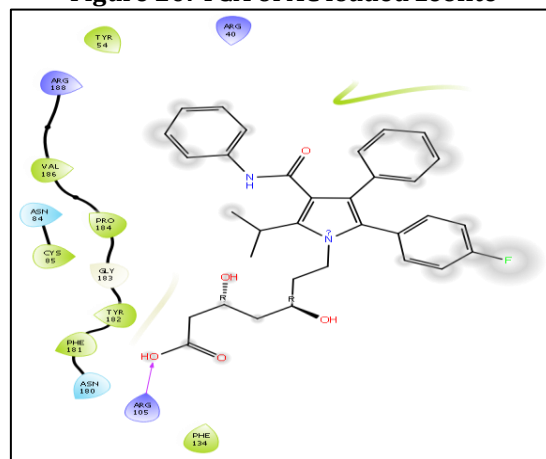


Figure 21: Docking site of Atorvastatin Calcium with receptor SARS CoV-2

Sr No	Parameters	Before Stability Study			After Stability Study		
		AC	MS	NU	AC	MS	NU
1	Appearance	Powder	Powder	Powder	Powder	Powder	Powder
2	Average Weight (mg)	100	100	100	99.5	99.2	99.1
3	Moisture Content (%)	4.50	5.00	4.74	4.45	4.90	4.40

**Figure 22: Stability study of drug loaded zeolite at storage condition of 40°C±2°C/75%RH±5%RH**

#### REFERENCES

- [1] Zahra AA, Giovanna S, Paolo B & M. Gabriella. Curcumin-loaded zeolite as anticancer drug carrier. *Pure Appl Chem* 2020; 92(3): 461–471.
- [2] Morteza S, Payam Z, Ghader M, Seok-Jhin K, Mohammad RG, Mohammad RS & Masoud M. Zeolites in drug delivery: Progress, challenges and opportunities. *Drug Discovery Today* 2020; 25(4): 642-656.
- [3] Indian Pharmacopoeia 2022; Published by Indian Pharmacopoeia Commission Ghaziabad; Volume-II & III: 1535-1537, 2913-2914 & 3013-3015.
- [4] British Pharmacopoeia 2019; Published by British Pharmacopoeia Commission London; Volume-I & II: 218-220, 288-289 & 347-348.
- [5] United States Pharmacopoeia 2019; Published by United States Pharmacopoeial Convention; North Bethesda, Maryland, United States; Volume-I & II: 410-417, 2879-2883 & 3023-3024.
- [6] KD Tripathi. *Essential of Medical Pharmacology*. Jaypee Brothers Medical Publishers, New Delhi, Sixth Edition (2008): 136-141, 184-196, 614-615.
- [7] RC Rowe, PJ Sheskey & ME Quinn. *Handbook of Pharmaceutical Excipients*. RPS Publishing, London, Sixth Edition (2009): 129-133, 393-396, 404-407, 506-509.
- [8] <https://pubchem.ncbi.nlm.nih.gov/compound/60822>
- [9] <https://go.drugbank.com/drugs/DB00461>
- [10] <https://www.epa.gov/assessing-and-managingchemicals-under-tsca/fact-sheet-methylene-chloride-ordichloromethane-dcm-0>
- [11] <https://www.sigmaaldrich.com/IN/en/product/sigald/688363>
- [12] Yudi W, Budipratiwi W and Tri AS. Enhancement of solubility and dissolution rate of atorvastatin calcium by cocrystallization. *Tropical Journal of Pharmaceutical Research* 2017; 16 (7): 1497-1502.
- [13] Sandhya M, Vishwanadham Y & Umema NT. Formulation And Evaluation of Atorvastatin Calcium Sustained Release Tablets. *Int J Pharm* 2016; 6(3): 124-130.
- [14] Moreshwar NK, Rajeshwar VK and Dinesh MS. Development And Validation of Spectrophotometric Method for Determination of Metoprolol Succinate. *Int J Chem Tech Res* 2009; 1(4): 1273-1277.
- [15] Rote AR and Bhalerao SR. Development and Statistical Validation of Spectrophotometric Methods for the Estimation of Nabumetone in Tablet Dosage Form. *E-Journal of Chemistry* 2010; 7(4): 1463-1467.
- [16] Spatarelu et al. Composite Nanogels Based on Zeolite-Poly (ethylene glycol) Diacrylate for Controlled Drug Delivery. *Nanomaterials* 2020; 10(2), 195.
- [17] Amorim R. et al. Zeolite Structures Loading with an Anticancer Compound As Drug Delivery Systems. *The Journal of Physical Chemistry C* 2012; 116(48): 25642–25650.

- [18] E. Kontogiannidou et al. In vitro and ex vivo assessment of microporous Faujasite zeolite (NaX-FAU) as a carrier for the oral delivery of danazol. *Journal of Drug Delivery Science and Technology* 2019; 51: 177–184.
- [19] Divband B, Rashidi MR, Khatamian M, Kazemi EGR, Gharehaghaji N and Dabaghi TF. Linde Type A and nano magnetite/NaA zeolites: cytotoxicity and doxorubicin loading efficiency. *Open Chemistry* 2018; 16(1): 21–28.
- [20] Dongban D, Hui L, Mengxin X, Mengqi C, Yuxiang H, Yaxin S & Zhibo L. Size-Controlled Synthesis of Drug Loaded Zeolitic Imidazolate Framework in Aqueous Solution and Size Impact on Their Cancer Theranostics In Vivo. *ACS Applied Materials & Interfaces* 2018;10(49): 42165–42174.
- [21] Krajis ND, Dakovic A, Malenovic A, Kragovic M & Milic J. Ibuprofen sorption and release by modified natural zeolites as prospective drug carriers. *Clay Minerals* 2015; 50(1): 11–22.
- [22] Patricia H, Carlos MA, Ainhua R, Joaquín PP & María VR. Controlled release of Ibuprofen from dealuminated faujasites. *Solid State Sciences* 2006;8: 1459–1465.
- [23] Chen WY, Jiao XJ, Liu YF, Liu JX & Zhang BH. Effect of 4A zeolite on adsorption and release properties of sodium alginate/ chitosan microspheres to sodium valproate. *Digest Journal of Nanomaterials and Biostructures* 2022; 17(3): 961 -977.
- [24] Dizaji et al. Synthesis of PLGA/chitosan/zeolites and PLGA/chitosan/metal organic frameworks nanofibers for targeted delivery of Paclitaxel toward prostate cancer cells death. *International Journal of Biological Macromolecules* 2020; 164: 1461-1674.
- [25] Mohammad K, Masoomeh A & Akbar M. Removal of Pb (II) and Cu (II) from aqueous solutions by NaA zeolite coated magnetic nanoparticles and optimization of method using experimental design. *Microporous and Mesoporous Materials* 2017;248: 256-265.
- [26] Qunpeng C, Hongxia L, Yilu X, Song C, Yuhua L, Fang D & Jianfen L. Study on the adsorption of nitrogen and phosphorus from biogas slurry by NaCl modified zeolite. *PLoS One* 2017; 12(5).
- [27] Ying W, Tanja EP, Krijn PJ and Jovana Z. Tailoring and visualizing the pore architecture of hierarchical zeolites. *Chem Soc Rev* 2015; 44: 7234-7261.
- [28] C. Karavasili et al. Comparison of different zeolite framework types as carriers for the oral delivery of the poorly soluble drug indomethacin. *International Journal of Pharmaceutics* 2017;528: 76–87.
- [29] Elham K, Reza H, Mona A, Roxana RB, Farzin H & Gholamhossein Z. Evaluation of synthetic zeolites as oral delivery vehicle for anti-inflammatory drugs. *Iran J Basic Med Sci* 2014; 17(5): 337–343.
- [30] Karimi M, Habibzadeh M, Rostamizadeh K, Khatamian M & Divband B. Preparation and characterization of nanocomposites based on different zeolite frameworks as carriers for anticancer drug: zeolite Y versus ZSM-5. *Polymer Bulletin* 2018; 76: 2233–2252.
- [31] Maharani DK, Sanjaya IGM, Amaria A, Anggraeni MA and Jannah LR. Molecular Docking Analysis Chitosan-Zeolite-ZnO Nanocomposite and Its Potency Against SARS-CoV-2. *IOP Conf. Ser.: Mater. Sci. Eng.* 2020; 1125-012006: 1-5.
- [32] <https://www.ich.org/page/quality-guidelines>
- [33] Chauhan SK, Tyagi A, Singh B, and Agarwal S. Accelerated Stability Studies of A Polyherbal Preparation (Eazmovr) Capsule. *Ancient Science of Life*; January & April 1999;18(3&4): 210-217.
- [34] Manikanta K, Deepthi R and Srinivasa RY. In Process Quality Control: Review. *World Journal of Pharmaceutical Research* 10(5): 1145-1158.
- [35] Hai-Xia Z, Jie-Xin W, Zhi-Bing Z, Yuan L, Zhi-Gang S & Jian-Feng C. Micronization of atorvastatin calcium by antisolvent precipitation process. *International Journal of Pharmaceutics*. 2009, 374(1–2): 106-113.
- [36] Praneeth KS, Suresh B, Raju J and Prabhakar RV. Formulation and Characterization of Floating Gelucire Matrices of Metoprolol Succinate. *Dissolution Technologies*; August 2010, 34-39.
- [37] Raj K, Ashutosh S, Kajal S, Divya D, Neha G & Prem FS. Preparation, characterization and in-vitro cytotoxicity of Fenofibrate and Nabumetone loaded solid lipid nanoparticles. *Materials Science and Engineering C* 2020;106:110184.
- [38] Shahnaz K, Mirzaagha B, Ali M, Parvin G & Moosa E. Preparation of hydroxyapatite-calcium ferrite composite for application in loading and sustainable release of amoxicillin: Optimization and modeling of the process by response surface methodology and artificial neural network. *Ceramics International* 2021;47(17): 24287-24295.
- [39] Mina K, Mahyar M, Abbasali Z & Bo L. Adsorption of phenol on environmentally friendly Fe<sub>3</sub>O<sub>4</sub>/chitosan/ zeolitic imidazolate framework-8 nanocomposite: Optimization by experimental design methodology. *Journal of Molecular Liquids* 2021;323:115064.

- [40] Altoom et al. Characterization of  $\beta$ cyclodextrin/phillipsite ( $\beta$ -CD/Ph) composite as a potential carrier for oxaliplatin as therapy for colorectal cancer; loading, release, and cytotoxicity. *Colloids and Surfaces A: Physicochemical and Engineering Aspects* 2022;648:129144.
- [41] Shamima NS, Ashima A and Mohiuddin AB. Design, Formulation and In Vitro Evaluation of Sustained Release Pulsatile Capsule of Metoprolol Tartrate. *IJPSR* 2015; 6(7): 2755-2761.
- [42] Senthilvel et al. Development of Atorvastatin Calcium Biloaded Capsules for Oral Administration of Hypercholesterolemia. *Evidence-Based Complementary and Alternative Medicine* 2022: 4995508.
- [43] Chirravuri SPK, Suma D, Lalitha M, Ashok T and Baskar NR. Formulation of Nabumetone Controlled Release Tablets Using Hpmc K4M. *Journal of Applicable Chemistry* 2016;5(1): 291-298.
- [44] Jayanthi B, Manna PK, Madhusudhan S, Mohanta GP and Manavalan R. Per oral extended release products –An overview. *Journal of Applied Pharmaceutical Science* 2011;1(2): 50-55.
- [45] Kawish et al. Development of nabumetone loaded lipid nano-scaffold for the effective oral delivery; optimization, characterization, drug release and pharmacodynamic study. *Journal of Molecular Liquids* 2017;231: 514-522.
- [46] Deshmukh VN, Singh SP & Sakarkar DM. Formulation and Evaluation of Sustained Release Metoprolol Succinate Tablet using Hydrophilic gums as Release modifiers. *Int J. PharmTech Res* 2009;1(2): 159-163.
- [47] Subal CB, Kesevan SK & Murugesan R. Design and release characteristics of sustained release tablet containing metformin HCl. *Brazilian Journal of Pharmaceutical Sciences* 2008;44(3): 477-483.
- [48] Richard WK, Robert G, Eric D, Pierre B & Nikolaos AP. Mechanisms of solute release from porous hydrophilic polymers. *International Journal of Pharmaceutics* 1983;15(1): 25-35.
- [49] Higuchi T. Mechanism of sustained-action medication. Theoretical analysis of rate of release of solid drugs dispersed in solid matrices. *Journal of Pharmaceutical Sciences* 1963; 52(12): 1145-1149.