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Flavonoids from *Pulicaria jaubertii* (Asteraceae) from Yemen.

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ABSTRACT

A phytochemical study of the *Pulicaria jaubertii* (Asteraceae) ethyl acetate extract resulted in the isolation, purification and characterization of two dihydroflavonols: dihydroquercetin 4'-methyl ether (1) Taxifolin (3) and one flavonol Quercetin 3 methyl ether (2). The structures were assigned on the basis of UV spectrophotometry as well as ¹H and ¹³C NMR spectroscopy.

Keywords: Asteraceae, *Pulicaria jaubertii*, dihydroflavonols, flavonoids.

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INTRODUCTION

Genus *Pulicaria* belonging to the tribe Inuleae of the Asteraceae family consists of 100 species distributed in Europe, North Africa and Asia and five species of this genus reported from Yemen [1]. Previous phytochemical investigations on several *Pulicaria* species have led to the identification of sesquiterpene lactones [2-3], diterpenes [4] as well as flavonoids [5-6]. Some of these substances such as flavonoids have exhibited many biological activities: antifungal [7], antimicrobial [8], antibacterial [9], antioxidant [10-11], cytotoxic [12] and anti-inflammatory [13] activities.

The antioxidant activity of CHCl_3 , EtOAc and *n*-butanol extracts of the aerial parts of *Pulicaria jaubertii* [14], as well as the chemical composition of its essential oil [15] have been reported previously. In continuation of our study on species of medicinal plants of the Asteraceae family, here we report the isolation and the structure elucidation of flavonol and dihydroflavonol from ethyl acetate extract of the soluble parts of the aqueous EtOH extract of the leaves of this species. The structures of the isolated compounds were identified on the basis of spectroscopic studies and comparison with literature values.

MATERIALS AND METHODS

Plant material

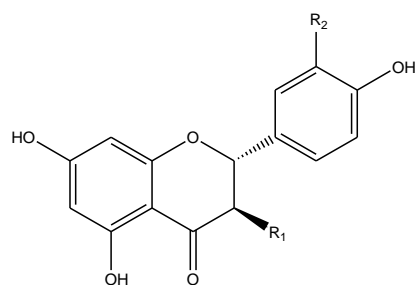
The leaves of *Pulicaria jaubertii* were collected from Aljar region (Hajjah - Yemen) in October 2008. The plant was identified by Pr. Abdellah Amine (Sana'a University). A voucher specimen of the plant material has been deposited at the department of biology (Sana'a University).

Extraction and isolation

Dried and powdered leaves (400 g) of *P. jaubertii* were extracted with 70% MeOH solution three times during 24 hours. The aqueous MeOH extract was concentrated to dryness. The residue was dissolved in water (200 ml). The resulting solution was extracted successively with petroleum ether, CHCl_3 , EtOAc and *n*-BuOH. The organic layers were dried with Na_2SO_4 to give after concentration the CHCl_3 (2 g), EtOAc (2.5 g) and *n*-BuOH (6 g) extracts respectively.

A part of the EtOAc extract (1 g) was chromatographed on preparative plates of silica gel eluted with CHCl_3 / MeOH (8.5:0.5) to yield 5 fractions (1–5). Fraction 3 (170 mg) was resubmitted to preparative TLC (CHCl_3 /MeOH; 9:1) to afford 7 sub-fractions (3-1; 3-7). The sub-fraction 3-2 was purified using Sephadex LH-20 to yield compounds **1**. Fraction 4 was subjected to repeated chromatography on silica gel by thin layer chromatography eluting with (CHCl_3 /MeOH; 9:1) to offered compounds **2** (8 mg) and **3** (8.2 mg).

RESULTS AND DISCUSSION



Compound	R ₁	R ₂	C ₂ -C ₃
1 Dihydroquercetin 4' methyl ether	OH	OCH ₃	Single
2 Quercetin 3 methyl ether	OCH ₃	OH	Double
3 Taxifolin	OH	OH	Single

Figure 1: Structures of identified compounds

All compounds (**1-3**) (Fig. 1) were isolated as yellow amorphous powder from the ethyl acetate extract. The UV spectra in methanol of compounds **1** and **3** showed similar behavior with maxima of absorbance at λ_{max} (nm) 289, 328 and 379 (sh) corresponding to $\pi \rightarrow \pi^*$ and $n \rightarrow \pi^*$ transitions, that matched the dihydroflavonol skeletons [16-17]. The ^1H NMR spectrum of these compounds exhibited a typical AX system due to H-2 and H-3 of a dihydroflavonol [18] at 4.85 (*d*, *J* 11.6 Hz) and δ H 4.44 (*d*, *J* 11.6 Hz) respectively. These assignments were confirmed by the ^{13}C NMR spectrum which showed three C-ring carbon signals at δ C85.5 (C-

2), 74.1 (C-3) and 198.5 (C-4) [19]. The trans-configuration of the upper dihydroflavonols unit could be deduced from the coupling constant between H₂ and H₃ ($^3J_{H_2-H_3}=11.6$ Hz) in the 1H NMR spectrum.

Compound 1: 1H NMR (CD₃OD, 400 MHz) δ (ppm) : 6.87 (1H, d, $J = 2.0$ Hz, H - 2'), 6.86 (1H, dd, $J = 8.0, 2.0$ Hz, H - 6'), 6.70 (1H, d, $J = 8.0$ Hz, H - 5'), 5.99 (1H, d, $J = 2.0$ Hz, H - 8), 5.95 (1H, d, $J = 2.0$ Hz, H - 6), 4.44 (1H, d, $J = 11.6$ Hz, H - 3), 4.85 (1H, d, $J = 11.6$ Hz, H - 2), 3.71 (3H, s, OCH₃). This compound was identified as 3',5,7-trihydroxy 4'-methoxy dihydroflavonol or dihydroquercetin 4'-methyl ether [6, 20].

Compound 2: UV Spectral Data, λ max (nm); MeOH: 256 nm, 358 nm, 294 nm; + NaOH: 270 nm, 407nm, 331 nm; + AlCl₃: 264 nm, 396 nm, 300 nm; + AlCl₃/HCl: 264 nm, 360 nm; + NaOAc: 264 nm, 369 nm; + NaOAc/H₃BO₃: 262 nm, 380 nm, 297 nm. 1H NMR (CD₃OD, 400 MHz) δ (ppm) : 7.42 (1H, sl, H - 2'), 7.35 (1H, dl, $J = 8.6$, Hz, H - 6'), 6.80 (1H, d, $J = 8.6$ Hz, H - 5'), 6.23 (1H, d, $J = 2.0$ Hz, H - 8), 5.95 (1H, d, $J = 2.0$ Hz, H - 6), 4.44 (1H, sl, H - 8), 6.05 (1H, sl, H - 6), 3.67 (3H, OCH₃). This compound was identified as Quercetin 3 methyl ether [21].

Compound 4: 1H NMR (CD₃OD, 400 MHz) δ (ppm) : 6.86 (1H, d, $J = 1.7$ Hz, H - 2'), 6.74 (1H, dd, $J = 1.7, 8.1$ Hz, H - 6'), 6.70 (1H, d, $J = 8.1$ Hz, H - 5'), 5.80 (1H, d, $J = 1.7$ Hz, H - 8), 5.76 (1H, d, $J = 1.7$ Hz, H - 6), 4.39 (1H, d, $J = 11.5$ Hz, H - 3), (H - 2 Covered with a signal Solvent). ^{13}C NMR (CD₃OD, 100 MHz) δ (ppm) : 85.47 (C - 2), 74.06 (C - 3), 198.51 (C - 4), 164.88 (C - 5), 98.01 (C - 6), 170.08 (C - 7), 97.01 (C - 8), 165.72 (C - 9), 102.01 (C - 10), 130.35 (C - 1'), 121.34 (C - 2'), 146.71 (C - 3'), 147.52 (C - 4'), 116.54 (C - 5'), 116.32 (C - 6'). This compound was identified as Taxifolin [22].

CONCLUSION

The present study allowed the isolation and structural determination of two dihydroflavonols named: dihydroquercetin 4'-methyl ether (1) and Taxifolin (3) together with substituted flavonol Quercetin 3 methyl ether (2). In our Knowledge, this is the first report describing the isolation of all these compounds from *P. jaubertii*.

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