

Research Journal of Pharmaceutical, Biological and Chemical Sciences

Synthesis, Characterisation and Antimicrobial Activity of Mannich Bases Derived From Benzhydrazide

A Abdul Jameel^a, M Syed Ali Padusha^{a*}, S Rathakrishnan^b, S Arun Prabhu^c

^aP.G & Research Department of Chemistry, Jamal Mohamed College, Tiruchirapalli-620 024, Tamilnadu, India

^bDepartment of Chemistry, SRR Engineering College, Chennai-603 103, Tamilnadu, India

^cDepartment of chemistry, National college, Tiruchirapalli-620 001, Tamilnadu, India

ABSTRACT

The present study deals with the synthesis, structure and antimicrobial properties of Mannich bases derived from Benzhydrazide. The structures of the synthesized Mannich bases were characterized by spectroscopic methods (IR, H^1 -NMR & ^{13}C -NMR) and analytical techniques (elemental analysis, melting point and TLC). The antimicrobial activities of synthesized compounds were also tested against certain organisms (*Staphylococcus aureus*, *P.aeruginosa*, *Bacillus spp.* and *Escherichia coli*).

Keywords: Benzhydrazide, Mannich base, Antimicrobial activities.

**Corresponding author*

E-mail: padusha_chem@yahoo.co.in

INTRODUCTION

Mannich bases have gained importance due to their applications in pharmaceutical chemistry [1,2] and they have several biological activities such as cytotoxic [3], local anaesthetic [4], antimicrobial [5], antimalarial [6], antihypertensive [7], antipsychotic, antitubercular [8], and antibacterial [9], activities. Further it has been observed from the literature that there has been a growing interest in the chemistry of aroylhydrazones[10]. Keeping in view of these observations and in continuation of our search for biologically active amino benzylated [11] Mannich bases, we planned to synthesized novel amino benzylated Mannich bases. Several reports are available for the synthesis of Mannich bases using 1^o and 2^o amines, aliphatic and aromatic aldehydes and compounds containing active hydrogen atom.

Many compounds such as acetone, substituted acetone, amides like urea, acetamide and semicarbazide were employed as compounds possessing active hydrogen atom but no report is available using benzhydrazide as active hydrogen compound for the synthesis of Mannich base. In the present study, Mannich bases have been prepared by reacting anisaldehyde/ benzaldehyde and benzhydrazide as fixed compounds and varying the amines as diethyl amine, morpholine and N-methyl piperazine.

We thereby report a set of new Mannich bases (I to VI) formed by the three component condensation reaction. The structures of the synthesized compounds have been established on the basis of FTIR, ¹H NMR and ¹³C NMR. The compounds have been screened for the *in vitro* antibacterial activities.

MATERIALS AND METHODS

Experimental

TLC was used to monitor the reactions and purity of the synthesized compounds. The melting points were determined in open capillary tubes and are uncorrected. IR Spectra were recorded in KBr disc on a Shimadzu IR Affinity 1, and ¹H NMR & ¹³C NMR spectra were measured on a Bruker Amx 400 NMR spectrometer using TMS as an internal standard. All Chemical shifts were reported on δ scales. The analytical data of all the compounds are presented in Table -2 and spectral data of all the compounds in

Synthesis of compound I – III

To an ethanolic solution of Benzhydrazide (0.025mol, 3.4 g) and Benzaldehyde (0.025mol, 2.65 ml), the amine (Diethyl amine, Morpholine and N-methyl piperazine) (0.025mol) was added in drops. The reaction mixture was kept on a magnetic stirrer and stirred well at room temperature. The product formed was washed with water and it was recrystallised with ethanol.

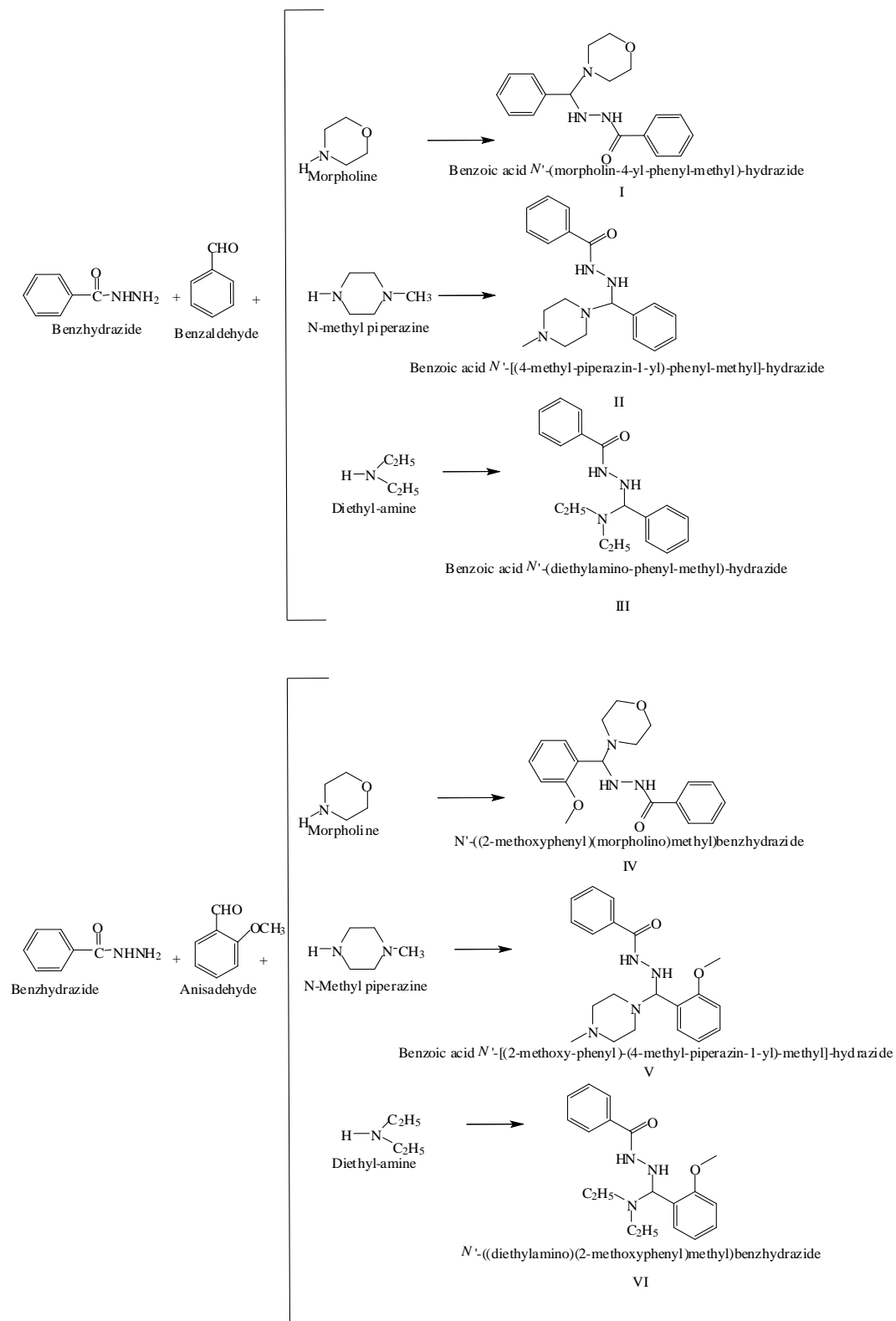


Table -1 Spectral data of compounds I – VI

Compd	IR (cm ⁻¹)	¹ H NMR	¹³ C NMR
I	3123 (N-H), 2839 (C-H), 1641 (C=O)	δ: 2.37 (m,4H,N-CH ₂), 3.67(m,4H,O-CH ₂), 5.04 (d,1H,CH), 7.14 (m,5H, Ar), 8.0 (d,2H,NH), 7.95 (m,5H,Ar)	δ: 114.129 (Ar), 51.3 (N-CH ₂), 77.8 (N CH – N), 164.9 (C=O)
II	3156 (N-H), 2840(C-H), 1630. (C=O)	δ: 2.27 (s,3H, N-CH ₃), 2.46 (m,8H, CH ₂), 7.95 (m,5H,Ar), 8.0 (d,2H,NH), 5.04 (d,1H,CH), 7.14 (m,4H,Ar)	δ: 127.5 (Ar), 167.3(C=O), 38.7 (N-CH ₃), 50.0 (N-CH ₂), 77.5 (N-CH-N)
III	3223 (N-H),2923 (C-H), 1641 (C=O)	δ: 1.02 (t,6H,CH ₃), 7.9(m,5H,Ar), 2.0(d,1H,NH), 8.0(d,1H,NH), 5.04(d,1H,CH), 7.14(m,5H,Ar), 2.40 (m,4H, CH ₂)	δ: 128.0 (Ar), 167.3 (C=O), 77.6 (CH), 13.7 (CH ₃), 41.2 (CH ₂)
IV	3128 (N-H),3003 (C-H), 1659 (C=O)	δ: 2.37 (m,4H,CH ₂), 3.67 (m, 4H,CH ₂), 3.73 (s,3H,CH ₃), 5.04 (d,1H,CH), 6.9 (m,4H, Ar), 8.0 (d,2H,NH), 7.44 (m,5H,Ar)	δ: 56.2 (O-CH ₃), 114.129 (Ar), 48.66 (N-CH ₂), 67.6 (N-CH – N), 164.9 (C=O)
V	3176 (N-H), 3010 (C-H), 1648 (C=O)	δ: 2.27 (s,3H, N-CH ₃), 3.73(s,3H,O-CH ₃), 2.46 (m,8H, CH ₂), 7.95 (m,5H,Ar), 8.0 (d,2H,NH), 5.04 (d,1H,CH), 6.95 (m,4H,Ar)	δ: 127.132 (Ar), 164.9 (C=O), 43.1 (N-CH ₃), 56.2 (O-CH ₃), 47.55 (N-CH ₂), 67 (N-CH-N)
VI	3123 (N-H),2939 (C-H) , 1641 (C=O)	δ: 1.0 (t,6H,CH ₃),7.9 (m,5H,Ar), 2.0 (d,1H,NH),5.04 (d,1H,CH),3.73 (s,3H,CH ₃),6.94(m,4H,Ar), 2.40 (m,4H, CH ₂),7.70(m,5H,Ar)	δ: 128.0 (Ar), 164.9 (C=O), 67.6 (CH), 56.2 (CH ₃), 13.4 (CH ₃), 44.0 (CH ₂)

Table- 2 Analytical Data

Compd	Yield (%)	m.w	m.p	M.F	Elemental analysis. Calculated (found) (%)		
					C	H	N
I	70	311	198	C ₁₈ H ₂₁ N ₃ O ₂	69.43(69.50)	6.80(6.75)	13.49(13.54)
II	68	324	220	C ₁₉ H ₂₄ N ₄ O	70.34(70.30)	7.46(7.50)	17.27(17.18)
III	73	297	210	C ₁₈ H ₂₃ N ₃ O	72.70(72.74)	7.80(7.70)	14.13(14.08)
IV	70	341	160	C ₁₉ H ₂₃ N ₃ O ₃	66.84(66.78)	6.79(6.70)	12.31(12.39)
V	72	354	177	C ₂₀ H ₂₆ N ₄ O ₂	67.77(67.81)	7.39(7.43)	15.81(15.72)
VI	68	327	182	C ₁₉ H ₂₅ N ₃ O ₂	69.70(69.65)	7.70(7.60)	12.83(12.73)

Table-3 Antibacterial activity of compounds I-VI.

Test Organisms	Zone of Inhibition (mm)						
	I	II	III	IV	V	VI	Ciprofloxacin
<i>Escherichia coli</i>	20	16	25	27	36	30	27
<i>p.aeruginosa</i>	13	22	15	22	16	19	20
<i>Bacillus spp.</i>	14	14	18	15	15	18	20
<i>Staphylococcus Aureus</i>	8	12	15	26	10	20	22

Synthesis of compound IV – VI

An ethanolic solution of Benzhydrazide, Anisaldehyde and amine (Diethyl amine, Morpholine and N-methyl piperazine) were taken in 1:1:1 mole ratio. Benzhydrazide (3.4 g, 0.025 mol), amine (Diethyl amine, Morpholine and N-methyl piperazine) (0.025 mol) and Anisaldehyde (3.4 ml, 0.025 mol) were mixed under ice-cold condition. The reaction mixture

was kept on a magnetic stirrer and stirred well. The product formed was washed with water and it was recrystallised with ethanol.

Antimicrobial activity

The synthesized compounds (I-VI) were screened for their antimicrobial activities against different strains of gram positive and gram negative bacteria's by disc diffusion method at concentration of 10 µg per ml in DMSO. The zone of inhibition was measured in mm. All the compounds possess appreciable activities against selected organisms. The zone of inhibition values are presented in the Table-3

CONCLUSION

A new series of Mannich base derivatives I-VI were synthesized and screened for antimicrobial activity. Among these, compounds **V** and **VI** have high activity against *Escherichia coli*, compound **II** and **IV** have high activity against *P.aeruginosa* and compound **IV** has high activity against *Staphylococcus aureus* compared with standard Ciprofloxacin, which can be beneficial for further studies. These synthesized compounds could be extended to analyze their various pharmacological activities.

REFERENCES

- [1] Pandeya SN, Lakshmi VS and Aandey A. Indian J Pharma Sci 2003; 65: 213.
- [2] Holla BS, Mahalinga M, Poojary B, Akbarali PM and Shetty NS. Indian J Heterocyclic Chemistry 2004; 14: 63.
- [3] Reddy M Vijaya Bhaskar, Chung-Rensu, Chiou Wen-Fei, Nan-Hiu Yi, Chen Rosemary Yin Hwa, Kenneth FB, Lee Kuo – Hsiung, Wu Tian – Shung. Bioorg Med Chem 2008; 16: 7358.
- [4] Rajan Ra Kali, Jubie S, Gowramma B, and Suresh B. Asian J Chem 2008; 20; 5289.
- [5] Ali Mohammed Ashraf and Shaharyar Mohammed. Boorg Med Chem Lett 2009; 17: 3314.
- [6] Chipeleme Alex, Gut Jiri, Rosenthal J Philip and Chibale Kelly. Bioorg Med Chem m 2007; 15: 273.
- [7] Kuwabe T, Okuyama S, Hashimoto S, Japan, Kokai Tokyo Koho Jp, (10) 1998, 114, 755. Appl 1996; 266(96): 274.
- [8] Valarmathi R, Karpagam S, Jayaker B. Indian J Heterocyclic chem. 2000; 10: 75.
- [9] Joshi S, Khosala N, Khare D and Sharada R. Bioorg Med Chem Lett 2005; 15: 221.
- [10] Duggal HK and Aggarwal DV. Synth react Inorg Metal Org Chem 1988; 18: 87. Dutta RL and Hossain MD M. Indian J Chem 1984; 23: 130. Singh B, Singh RN and Aggarwal RC. Synth React Inorg Metal Org Chem 1984; 14.
- [11] Chaluvaraju KC and Ishwarbhat K. Asian J Chem 2008; 20: 4335.