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Studies of relationship between swelling and drug release in the sustained release hydrophilic matrices containing different grades of hydroxypropylmethylcellulose.

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ABSTRACT

The current study examines the relationship between swelling and drug release from the hydrophilic matrices of Chlorpheniramine maleate prepared using combination of different grades of hydroxypropylmethylcellulose (HPMC), viz, HPMCK4M, HPMCK15M and HPMCK100M. The results indicate that swelling and release profiles were affected by concentration and viscosity grade of the polymer. In order to elucidate the release mechanism, the data were fitted to equation described by Peppas and Korsmeyer ($Mt/M \alpha Kt^n$). The value of release rate exponent (n) is a function of geometric shape of the drug delivery device. The results indicate that the mechanism of release is influenced greatly by the polymer concentration of the formulations as can be seen from the r^2 values and n was generally in accordance with these indications. In this investigation it has been demonstrated that an inverse relationship exists between the drug release rate and matrix-swelling rate. When the amount of HPMC in the matrix is high, wetting improves and water uptake into matrices is enhanced. The higher amount of HPMC causes a greater degree of swelling this in turn reduces the drug release, as the diffusional path length of drug is now longer. Conversely, reduction in the amount of HPMC reduces the degree of swelling and the thickness of gel layer, this enables faster drug release. Swelling studies reveals an inverse relationship between swelling and drug release in the sustained release hydrophilic matrices containing different grades of the hydroxy**propylmethylcellulose.**

Key words: Hydroxypropylmethylcellulose, Hydrophilic matrices, swelling

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INTRODUCTION

Hydrophilic matrices devices are one of the least complicated approaches in the formulation of sustained release dosage forms and are finding increasing application in the pharmaceutical field. The aim of the present study was to investigate relationship between swelling and drug release from the sustained release hydrophilic matrices of Chlorpheniramine maleate prepared using combination of different grades of hydroxypropylmethylcellulose (HPMC), viz, HPMCK4M, HPMCK15M and HPMCK100M [1,2]. Drug release data from HPMC matrices follows the classical Higuchi dissolution equation, relating drug release with square root of time. Swellable systems consisting of hydrophilic polymers, in the presence of water, absorb a significant amount of water to form a gel. As the dissolution medium penetrates the matrix, polymer material swelling starts and drug molecules begin to move out of the system by diffusion. The degree of swelling and percent water uptake is determined to find the relationship between the drug release and swelling [3-8]. The release mechanism is obtained from the dissolution data and the value of release rate exponent is determined. The value of release rate exponent (n) is a function of geometric shape of the drug delivery device. The results indicate that the mechanism of release is influenced greatly by the polymer concentration of the formulations as can be seen from the r^2 values and n was generally in accordance with these indications. The release is mainly determined by the Fickian diffusion which is also confirmed from the n values [9-12].

MATERIALS AND METHODS

Chlorpheniramine maleate was obtained as a gift sample from Pure Pharma. Ltd, Indore, (M.P.), Methocel (K4M, K15M, K100M) were provided by Colorcon India Ltd., Goa, dicalcium phosphate, microcrystalline cellulose (Avicel pH_{101}), talc , magnesium stearate and all other reagent used were of analytical grade.

Preparation of Matrices

Nine formulations employed for investigations containing different ratios of HPMC of different grades were prepared by direct compression and coded C1, C2, C3, D1, D2, D3, E1, E2 and E3. The ratios of different grades of HPMC employed are shown in Table [1]. The amount of drug, magnesium stearate, MCC and talc were kept constant while dicalcium phosphate was taken in sufficient quantity to maintain a constant tablet weight of 120 mg. All the products and process variables (other than the concentrations of two polymers) like mixing time, compaction force, etc, were kept constant. Ten tablets from each batch were weighed individually and subjected to physical evaluation.

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Table 1: Different ratios employed in formulations containing HPMC of different grades

Formulation Code	HPMCK4M	HPMCK100M	CPM
C1	1	1	1
C2	2	2	1
C3	3	3	1
Formulation Code	HPMCK4M	HPMCK15M	CPM
D1	1	1	1
D2	2	2	1
D3	3	3	1
Formulation Code	HPMCK15M	HPMCK100M	CPM
E1	1	1	1
E2	2	2	1
E3	3	3	1

Matrix Swelling and Water Uptake Studies

Swelling was evaluated by weight. The matrices were placed in 900 ml dissolution medium pH 6.3, at 37^0C. At different time intervals, the previously weighed tablets were removed, gently wiped with a tissue to remove surface water, and reweighed. The percent water uptake i.e., degree of swelling due to absorbed test liquid, can be estimated at regular time intervals using the following equation –

$$\% \text{ water Uptake} = (Ws-Wi)/Wp *100$$

Where, Ws = Wt. of the swollen matrix at time t, Wi = Initial wt. of the matrix, Wp = wt. of the polymer in the matrix. The polymer swelling or water uptake are mean of three determinations. The degree of swelling can be calculated by the following formula –

$$\text{Degree of swelling} = Ws-Wd/Wd*100$$

Where, Wd = Final dry wt. of the matrix, Ws = Swollen wt. of the same matrix at immersion time (t). The swelling degree is the mean of at least three determinations.

Dissolution Studies

Dissolution studies were carried out for all the nine formulations in triplicate, employing dissolution apparatus, using distilled water pH 6.3 as the dissolution medium at 50 rpm and 37 $\pm$ 0.5^0C. An aliquot of sample was periodically withdrawn at suitable time intervals and volume replaced with equivalent amounts of plain dissolution medium. The samples were analyzed at 261 nm.

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RESULTS AND DISCUSSION

The weight of the polymer in the matrix (Wp) and final dry weight of the matrix (Wd) are shown in Table [2]. The weight of the swollen matrix at different time intervals, degree of swelling and percent water uptake data was observed and reported Table [3].The results of swelling studies are shown graphically in Double –Y plots showing dissolution profiles of Chlorpheniramine maleate release and swelling from matrices containing HPMC K4M and K100M grades combinations, (formulation codes C1, C2, C3, Fig. [1a].The percent uptake swelling or water uptake plots are shown in Fig. [1b]. Similar plots are shown in Fig [2a] and Fig [2b] for formulation codes D1, D2, D3, containing HPMC K4M and K15M combinations with different ratios and Fig [3a] and Fig [3b] for formulation codes E1, E2, E3, containing HPMC K15M and K100M combinations with different ratios. The dissolution parameters of varied formulation with different ratios of polymer combinations obtained during studies are shown in Table [3].

Table 2: Final dry weight and weight of polymer in matrix tablets of different batches

Formulation Code	Final Dry weight (Wd) (mg)	Weight of polymer in matrix (Wp) (mg)
C1	120	24
C2	127	48
C3	125	72
D1	121	24
D2	120	48
D3	124	72
E1	124	24
E2	125	48
E3	122	72

Table 3: Dissolution parameters, Degree of swelling and Percent water uptake

Formulation Code	Release at 12 hr	Release at 24 hr	n	Degree of Swelling (%)	Percent of water uptake	r^2
C1	96.54	N.C.	0.504	377.50	1887.50	0.986
C2	84.27	101.35	0.453	388.90	1029.20	0.979
C3	75.00	84.82	0.444	401.60	697.20	0.977
D1	104.00	N.C.	0.551	323.10	1212.50	0.974
D2	102.73	N.C.	0.547	345.00	722.90	0.975
D3	86.00	102.08	0.459	353.20	608.30	0.971
E1	93.05	N.C.	0.508	369.30	1737.50	0.968
E2	74.99	84.82	0.444	396.80	1033.30	0.977
E3	63.50	91.58	0.431	420.50	712.50	0.980

Formulation C1 has n= 0.504, C2 has n=0.453 and C3 has n=0.444 indicating that the release mechanism is very close to Fickian transport i.e. belong to the Higuchi model. In this investigation it has been demonstrated that an inverse relationship exists between

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the drug release rate and matrix-swelling rate. When the amount of HPMC in the matrix is high, wetting improves and water uptake into matrices is enhanced. The higher amount of HPMC causes a greater degree of swelling. This in turn reduces the drug release, as the diffusional path length of drug is now longer. Conversely, reduction in the amount of HPMC reduces the degree of swelling and the thickness of gel layer, this enables faster drug release. Similar results are observed with the different viscosity grades of HPMC formulations, viz D1, D2, D3 and E1, E2, E3. HPMC of higher viscosity grades swells to greater extent and has greater intrinsic water uptake property than that of the lower viscosity grades.

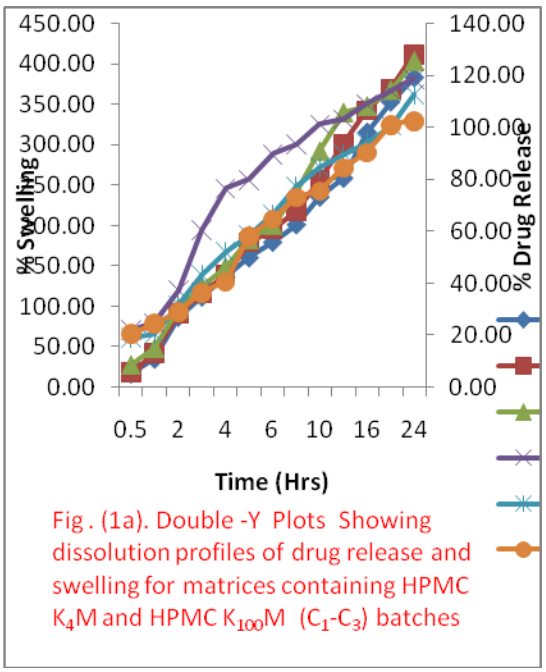

Fig . (1a). Double -Y Plots Showing dissolution profiles of drug release and swelling for matrices containing HPMC K_4M and HPMC $K_{100}M$ (C_1-C_3) batches

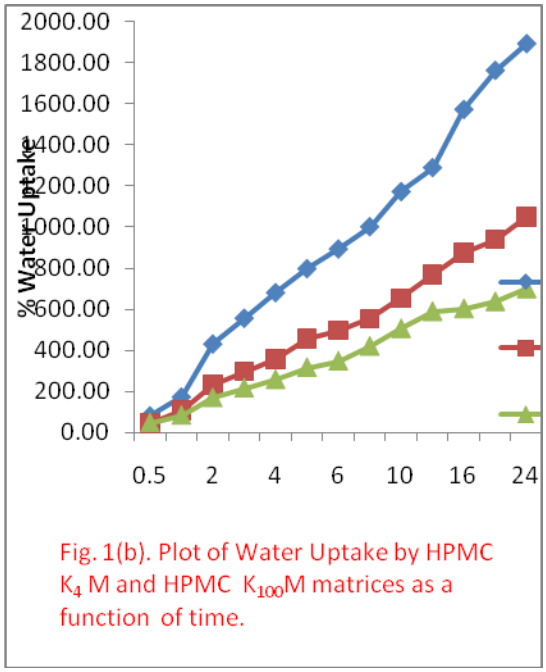

Fig. 1(b). Plot of Water Uptake by HPMC K_4 M and HPMC $K_{100}M$ matrices as a function of time.

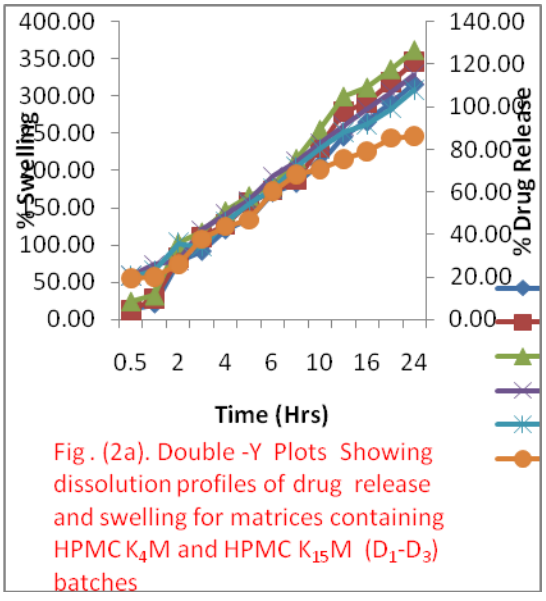

Fig . (2a). Double -Y Plots Showing dissolution profiles of drug release and swelling for matrices containing HPMC K_4M and HPMC $K_{15}M$ (D_1-D_3) batches

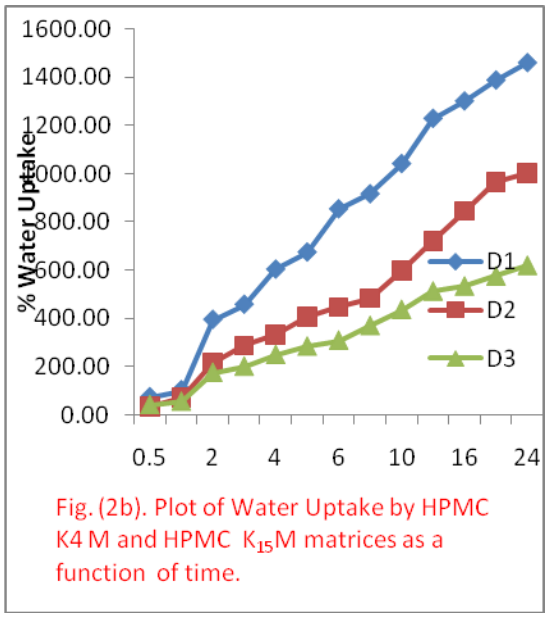

Fig. (2b). Plot of Water Uptake by HPMC K4 M and HPMC $K_{15}M$ matrices as a function of time.

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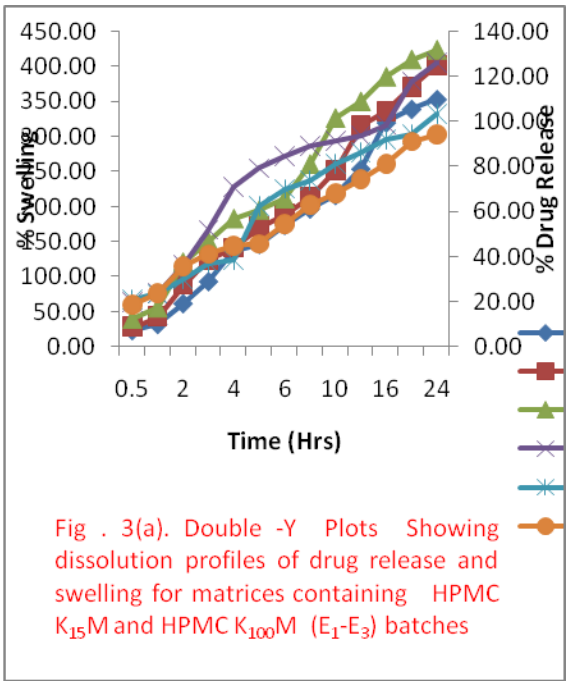

Fig . 3(a). Double -Y Plots Showing dissolution profiles of drug release and swelling for matrices containing HPMC K_{15}M and HPMC K_{100}M (E_1-E_3) batches

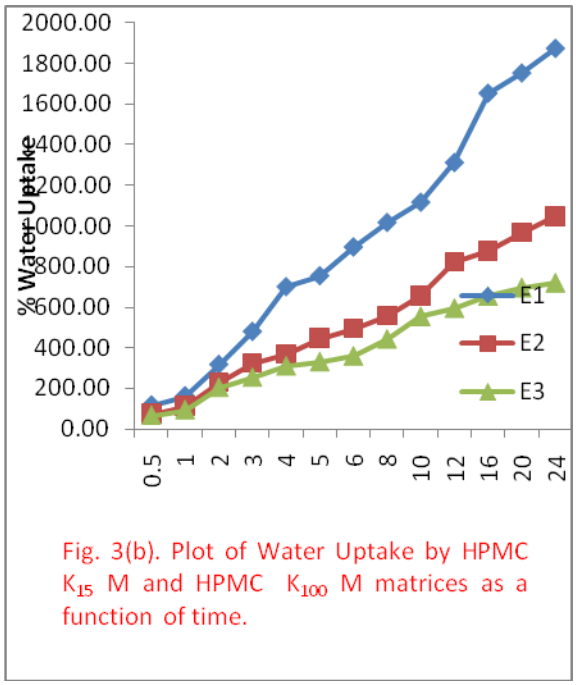

Fig. 3(b). Plot of Water Uptake by HPMC K_{15} M and HPMC K_{100} M matrices as a function of time.

CONCLUSION

Swelling studies reveals an inverse relationship between swelling and drug release. The rational combination of different grades of HPMC tends to provide quite regulated release of Chlorpheniramine maleate over an extended period of time.

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